

Solutions and Supplies

1. Phosphocellulose-purified tubulin (~50 mg = 2-4 of 3ml aliquots of 5-10 mg/ml PC fractions)
2. Dye stock in anhydrous DMSO (20-100 mM)
3. BRB80 (1X): 80 mM PIPES, 1 mM MgCl₂, 1 mM EGTA, pH 6.8 with KOH (generally made as a 5X stock and stored at 4°C)
4. High pH Cushion: 0.1 M NaHEPES, pH 8.6, 1 mM MgCl₂, 1 mM EGTA, 60% (v/v) glycerol
5. Labeling Buffer: 0.1M NaHEPES, pH 8.6, 1 mM MgCl₂, 1 mM EGTA, 40% (v/v) glycerol
6. Quench: 2X BRB80, 100 mM K-Glutamate, 40% (v/v) glycerol
7. Low pH cushion: 60% (v/v) glycerol in 1X BRB80
8. MLA80 rotor (warm = 37°C)
9. TLA100.4 or TLA100.3 and TLA100.2 rotors
10. Note: 1M HEPES, pH to 8.6 with NaOH and store at -20°C
11. 2M K-Glutamate - dissolve glutamic acid to 2M, carefully pH with KOH such that 50 mM has a pH ~7.0 and store at -20°C

All buffers for labeling can be stored indefinitely at -20°C; dye stocks are best prepared fresh from powder that has been stored anhydrously at -20°C; residual dye solution can be stored at -20°C or -80°C under anhydrous conditions.

Labeling Protocol

The procedure described below can be scaled down if desired. It is essential to perform all steps involving caged dyes under a safelight in a room well-shielded from light. A piece of red acetate sheet taped over a dimly lit lamp is adequate as a safelight. Other dye labelings can be done under room light, minimizing exposure during incubations by using foil.

A high pH is used in order to enhance yield in the dye coupling reaction.

1. Thaw 2 PC column fractions (30-60 mg tubulin) and add BRB80 to 0.5X, MgCl₂ to 3.5 mM, GTP to 1 mM and store on ice for 5'. Add half volume of glycerol for a final concentration of 33%.
2. Incubate for 1h at 37°C.
3. Layer polymerized tubulin onto 4 ml warm (37°C) High pH Cushion in 6 MLA80 tubes. Pellet microtubules in ultracentrifuge in a MLA80 rotor at 70K rpm for 35' at 35°C.
4. Aspirate the supernatant above the cushion and rinse the supernatant-cushion interface twice with warm (37°C) Labeling Buffer. Aspirate the cushion and resuspend the pellet using a cutoff large pipet tip in 0.5 ml of warm labeling buffer. Take care to keep the tubulin warm during the resuspension and continue resuspending till no chunks of tubulin are visible. This is the most painful part of the labeling procedure.
5. Add 10- to 20-fold molar excess of the dye to tubulin. Estimate the tubulin concentration assuming ~70% recovery of the starting tubulin. For dyes such as Cy5 and Cy3, use a 5-pack for labeling ~25 mg. For most dyes we label for 10' at 37°C. For C2CF-SNHS (caged fluorescein), we have found it best to add the dye in two steps (20' apart) and label for 60' at 37°C. After adding the dye stock, gently vortex the mixture every 2'-3' during the course of the labeling.
6. At end of labeling incubation add an equal volume of Quench to the labeling reaction and mix well. Incubate for 5'.

7. Layer the quenched labeling reaction onto two TLA100.3 (or TLA100.4) tubes containing 1.5 ml of Low pH Cushion. Spin at 80K for 20 min at 35°C in a TLA100.3 or TLA100.4 rotor in a Beckman TLA100 ultracentrifuge.
8. Aspirate the supernatant above the cushion and rinse the supernatant-cushion interface twice with warm 1XBRB80. Aspirate the cushion and resuspend the pellet using a cutoff pipet tip in 0.5 ml of ice-cold 1X BRB80.

At this point the Mitchison lab protocol recommends douncing the pellet to fully resuspend pellet as follows: Transfer resuspended chunks of the pellet to a small ice-cold dounce (1 or 2 ml volume) in an ice-water bath. Resuspend the pellet by gentle douncing till the suspension is uniform. Continue douncing intermittently for a total time of 30 min at 0°C.

However, Kazu, Tim, Fedor, Mark and myself (Jeff) have never had problems resuspending by pipet alone, and do not feel that the douncing optimizes this protocol at all. However, what does seem to be important is keeping the volume in this step low - this increases the overall tubulin concentration, which does improve the next polymerization step.

9. Spin the depolymerized tubulin in a TLA100.2 (or TLA100.3) rotor at 80K for 10' at 2°C.
10. Recover the supernatant from the cold spin, add BRB80 to 1X (from a 5X stock), MgCl₂ to 4 mM, GTP to 1 mM and incubate on ice for 3'. Warm to 37°C for 2', add 1/2 volume of glycerol (33% v/v final), mix well and polymerize at 37°C for 30 min.
11. Layer the polymerization reaction on a 1 ml Low pH Cushion in a TLA100.3 tube and pellet the microtubules at 80K in a TLA100.3 rotor for 20' at 37°C.
12. Aspirate the supernatant above the cushion and rinse the supernatant-cushion interface twice with warm 1X BRB80. Aspirate the cushion and rinse the pellet twice with 1 ml warm 1X BRB80 to remove any residual glycerol. Resuspend the pellet using a cutoff pipet tip in 0.2-0.3 ml of ice cold 1X BRB80. This pellet should resuspend easily. Incubate at 0°C for 20 to 30 min.
13. Spin the depolymerized tubulin in a TLA100 or TLA100.2 rotor at 80K for 10' at 2°C. Recover the supernatant, pool together, quickly estimate the tubulin concentration, adjust with 1X BRB80 if desired and freeze in 3 - 5 ul aliquots in liquid nitrogen. We generally aim for a final tubulin concentration of 5 - 15 mg/ml (50 - 150 μM). Careful determination of tubulin concentration and labeling stoichiometry can be performed as described below, after the tubulin has been aliquoted and frozen. C2CF-tubulin should be stored at -80°C in a foil-wrapped box.

Quantifying Tubulin Concentration and Labeling Stoichiometry

To determine the tubulin concentration and stoichiometry of labeling, dilute the labeled tubulin 1/50 - 1/100 in IB and obtain a wavelength spectrum. Calculate the molar concentration of dye by using the absorbance at the peak wavelength and the extinction coefficient provided by the dye manufacturer. Determine the tubulin concentration by first subtracting out the contribution of the dye to the A₂₈₀ and then using an extinction coefficient of 115,000 M⁻¹cm⁻¹. Section V provides a list of extinction coefficients and A₂₈₀ absorbance (relative to absorbance at peak wavelength) for commonly used dyes. Note that the absorbance of fluorescein is pH-dependent and conjugates with fluorescein should either be diluted into a high pH buffer (~8.8-9.0) or the value measured at pH 7.0 multiplied by 1.2.

An example of calculating concentration and stoichiometry for tubulin labeled with tetramethylrhodamine (TMR) NHS ester:

Labeling Stoichiometry = TMR concentration / Tubulin concentration

TMR concentration = (A₅₅₅ x Dilution Factor) / Extinction Coeff. of TMR at 555 nm

Tubulin concentration = (A₂₈₀ - Contribution of dye to A₂₈₀) x Dilution / Extinction coefficient of tubulin at 280 nm (=115000)

Example:

A wavelength spectrum of 1/100 dilution of the final labeled tubulin product gave the following absorbance values:

$A_{280} = 0.23$; $A_{555} = 0.20$. Therefore,

Tubulin concentration = $\{(0.23 - (0.2 \times 0.2)) \times 100\} / 115000 = 165 \mu\text{M}$

TMR concentration = $[0.20 \times 100] / 95000 = 210 \mu\text{M}$

Labeling Stoichiometry = $210 / 165 = 1.3$

To determine the concentration and labeling stoichiometry of C2CF-tubulin, the C2CF must be first uncaged to fluorescein. To do this, dilute the labeled tubulin 1/50 to 1/100 in IB + 2 mM DTT in an eppendorf tube. Put the eppendorf tube on a hand held UV lamp, cover with foil (shiny side down) and expose to long wavelength UV for 30'. Obtain a wavelength spectrum from 200 to 600 nm after the 30' activation, using IB + 2 mM DTT exposed to UV in parallel as a blank. Assuming a 100% efficiency for the uncaging reaction, the concentration of C2CF can be calculated from the spectrum after activation as follows:

Concentration of C2CF = $(A_{495} \times \text{Dilution Factor} \times 1.2) / 74000$

(The factor of 1.2 corrects for the pH dependence of the absorption spectrum of fluorescein)

Using Labeled Tubulins

1. Microinjection into cells/addition to extracts: For microinjections, we dilute the tubulin in IB to 2-5 mg/ml, clarify by centrifugation and inject ~1/10 of cell volume. For frog extract studies, we add labeled tubulin to 1/40 - 1/200th of the extract tubulin pool (~20 μM).
2. Preparation of fluorescent microtubule substrates or for monitoring polymerization and dynamics of pure tubulin: We use a mixture of labeled and unlabeled tubulin for polymerization. The ratio of labeled to unlabeled will depend on the particular application and on the brightness of the labeled tubulin. Labeled tubulins, especially those labeled to high stoichiometry, exhibit very different properties from unlabeled tubulin. Therefore, we use the highest ratio of unlabeled to labeled tubulin that provides signal intensity sufficient for a particular experiment.

Properties of Fluorescent Dyes Used for Tubulin Labeling

Dye	Excitation (nm)	Emission (nm)	$\epsilon_{\text{max}}(\text{M}^{-1}\text{cm}^{-1})$	$\epsilon_{280}/\epsilon_{\text{max}}$
Fluorescein	495	519	74,000	0.19
Oregon Green 488	495	521	76,000	0.19
Cy3	550	570	150,000	0.08
Tetramethylrhodamine	550	576	95,000	0.21
X-Rhodamine	574	602	78,000	0.20
Texas Red	583	603	116,000	0.15
Cy5	649	670	250,000	0.05

ϵ_{max} = Extinction coefficient of dye at its peak wavelength

$\epsilon_{280}/\epsilon_{\text{max}}$ = Absorbance of dye at 280 nm as a fraction of its absorbance at its peak wavelength

Some catalog numbers:

- 5 (and-6) carboxyfluorescein succinimidyl ester (Molecular Probes C-1311)
- Oregon Green 488 carboxylic acid, succinimidyl ester 5-isomer (Molecular Probes O-6147)
- 5 (and-6) carboxytetramethylrhodamine succinimidyl ester (Molecular Probes C-1171)

- 5 (and-6) carboxy-X-rhodamine succinimidyl ester (Molecular Probes C-1309)
- Texas Red-X succinimidyl ester, mixed isomers (Molecular Probes T-6134)
- Cy3-OSu monofunctional reactive fluorophore (Amersham PA13100)
- Cy5-OSu monofunctional reactive fluorophore (Amersham PA13600)